

Exercise 6.3c

Ionic Formula to Name - Answers

Name: _____

Date: _____ Per: _____

DIRECTIONS: Name the following compounds in the space provided.

1. KCl potassium chloride
2. NaOH sodium hydroxide
3. ZnF₂ zinc fluoride
4. CuNO₂ copper (I) nitrite
5. NaMnO₄ sodium permanganate
6. KHSO₃ potassium hydrogen sulfite
7. Mn(OH)₂ manganese (II) hydroxide
8. CrCl₃ chromium (III) chloride
9. (NH₄)₂SO₄ ammonium sulfate
10. AgCl silver chloride
11. PbCrO₄ lead (II) chromate
12. NaI sodium iodide
13. NiCl₂ nickel (II) chloride
14. Hg₂Cl₂ mercury (I) chloride
15. Cu(NO₃)₂ copper (II) nitrate
16. Al₂(C₂O₄)₃ aluminum oxalate
17. Na₂Cr₂O₇ sodium dichromate
18. AgClO₃ silver chlorate
19. KH₂PO₄ potassium dihydrogen phosphate
20. Cr₂S₃ chromium (III) sulfide
21. NaF sodium fluoride
22. Cu(NO₂)₂ copper (II) nitrite
23. FeBr₂ iron (II) bromide
24. HgCO₃ mercury (II) carbonate
25. NH₄OH ammonium hydroxide
26. CrCl₂ chromium (II) chloride
27. Sn(NO₃)₂ tin (II) nitrate
28. Sn(NO₂)₄ tin (IV) nitrite
29. KOH potassium hydroxide
30. Hg(OH)₂ mercury (II) hydroxide
31. RbCl rubidium chloride
32. FeCl₃ iron (III) chloride
33. Cu₂O copper (I) oxide
34. Al₂(SO₄)₃ aluminum sulfate
35. Pb(OH)₂ lead (II) hydroxide
36. NH₄NO₃ ammonium nitrate
37. NaHCO₃ sodium hydrogen carbonate
38. HgO mercury (II) oxide
39. Zn(NO₂)₂ zinc nitrite
40. CsOH cesium hydroxide
41. Li₂O lithium oxide
42. Ca(OH)₂ calcium hydroxide
43. CaBr₂ calcium bromide
44. Fe₂O₃ iron (III) oxide
45. FeCO₃ iron (II) carbonate
46. Ba(BrO₃)₂ barium bromate
47. Al(OH)₃ aluminum hydroxide
48. NaC₂H₃O₂ sodium acetate
49. Na₂SO₃ sodium sulfite
50. K₂CO₃ potassium carbonate
51. NH₄IO₃ ammonium iodate
52. LiH lithium hydride
53. MgBr₂ magnesium bromide
54. SnBr₂ tin (II) bromide
55. NH₄F ammonium fluoride
56. BiCl₃ bismuth (III) chloride
57. KHCO₃ potassium hydrogen carbonate
58. K₂O potassium oxide