

Exercise 6.3c

Ionic Formula to Name

Name: _____

Date: _____ Per: _____

DIRECTIONS: Name the following compounds in the space provided.

- | | |
|---|---|
| 1. KCl _____ | 30. Hg(OH) ₂ _____ |
| 2. NaOH _____ | 31. RbCl _____ |
| 3. ZnF ₂ _____ | 32. FeCl ₃ _____ |
| 4. CuNO ₂ _____ | 33. Cu ₂ O _____ |
| 5. NaMnO ₄ _____ | 34. Al ₂ (SO ₄) ₃ _____ |
| 6. KHSO ₃ _____ | 35. Pb(OH) ₂ _____ |
| 7. Mn(OH) ₂ _____ | 36. NH ₄ NO ₃ _____ |
| 8. CrCl ₃ _____ | 37. NaHCO ₃ _____ |
| 9. (NH ₄) ₂ SO ₄ _____ | 38. HgO _____ |
| 10. AgCl _____ | 39. Zn(NO ₂) ₂ _____ |
| 11. PbCrO ₄ _____ | 40. CsOH _____ |
| 12. NaI _____ | 41. Li ₂ O _____ |
| 13. NiCl ₂ _____ | 42. Ca(OH) ₂ _____ |
| 14. Hg ₂ Cl ₂ _____ | 43. CaBr ₂ _____ |
| 15. Cu(NO ₃) ₂ _____ | 44. Fe ₂ O ₃ _____ |
| 16. Al ₂ (C ₂ O ₄) ₃ _____ | 45. FeCO ₃ _____ |
| 17. Na ₂ Cr ₂ O ₇ _____ | 46. Ba(BrO ₃) ₂ _____ |
| 18. AgClO ₃ _____ | 47. Al(OH) ₃ _____ |
| 19. KH ₂ PO ₄ _____ | 48. NaC ₂ H ₃ O ₂ _____ |
| 20. Cr ₂ S ₃ _____ | 49. Na ₂ SO ₃ _____ |
| 21. NaF _____ | 50. K ₂ CO ₃ _____ |
| 22. Cu(NO ₂) ₂ _____ | 51. NH ₄ IO ₃ _____ |
| 23. FeBr ₂ _____ | 52. LiH _____ |
| 24. HgCO ₃ _____ | 53. MgBr ₂ _____ |
| 25. NH ₄ OH _____ | 54. SnBr ₂ _____ |
| 26. CrCl ₂ _____ | 55. NH ₄ F _____ |
| 27. Sn(NO ₃) ₂ _____ | 56. BiCl ₃ _____ |
| 28. Sn(NO ₂) ₄ _____ | 57. KHCO ₃ _____ |
| 29. KOH _____ | 58. K ₂ O _____ |